# Cyanine fluorophore derivatives with enhanced photostability

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Fluorescence applications requiring high photostability often depend on the use of solution additives to enhance fluorophore performance. Here we demonstrate that the direct or proximal conjugation of cyclooctatetraene (COT), 4-nitrobenzyl alcohol (NBA) or Trolox to the cyanine fluorophore Cy5 dramatically enhanced fluorophore photostability without otherwise affecting its native spectral characteristics. Such conjugation is a powerful means of improving the robustness of fluorescencebased applications demanding long-lived, nonblinking fluorescence emission.

#### BRIEF COMMUNICATIONS

#### Cvanine fluorophore derivatives with enhanced photostability

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Small organic fluorophores are powerful research tools in biological imaging that have enabled unprecedented insights into both cellular and molecular processes. However, their performence can be compromised by undesirable photophysical properties that limit both the fluorescence quantum yield and the total number of TSQ-fluorophore conjugates were largely indistinguishable from of photons emitted before photobleaching. Such issues include both transient (blinking) and irreversible (photobleaching) lightinduced transitions to dark states. Dark state transitions are nurticularly limiting in single-molecule studies that demand high by 29%, unchanged and decreased by 30%, respectively, compare illumination intensities. These problems are expecially common to that for CVS. Such changes more in part, reflect variations in the s longer-wavelength fluorophores, such as Cy5, widely used in storescence resonance energy transfer-based investigations and

powerful means of minimizing fluorophore blinking and photo-bleaching both in vitro and in vivo<sup>1.5</sup>. Common additives

using hydrophobic interaction chromatography (Online Methods)

Using wide-field, total internal reflection fluorescence (TIRF include triplet-state quenchers (TSQs) such as cyclooctatetraene imaging<sup>2</sup>, we assessed the photophysical properties of surface-(COT), 4-nitrobenzyl alcohol (NBA) and 6-hydroxy-2,5,7,8-immebilized fluorophore-DNA complexes at the single-molecule (COT), 4-nitrobenzyl alcohol (NBA) and 6-hydroxy-2,5,7,8termobilized fluorophore-DNA complexes at the single-molecule
tetramethylchromen-2-corboxylic acid (Troba) MM that act in a
scale under direct laser illumination at 640 nm. We used typical

Despite their advantages, TSQs have key limitations, including poor aqueous solubility, problems with membrane permeability and biological toxicity. To circumvent these issues, here we show that direct or proximal linkage of TSQs to the Cy5 fluorophore reduced blinking and photoblesching in both decaygemated and oxygenated environments to extents exceeding those using TSOs living cells, suggesting these new fluorophore derivatives may be

valuable for in vive applications'.

In vitro single-molecule studies demonstrating that TSQs open ate in a concentration-dependent fashion to affect the photophysical properties of cyanine fluorophores' suggest a collision-base mode of action<sup>6</sup>. To determine whether additional enhancement effective TSO concentration beyond the solubility limit while specific Cv5-TSQ conjugates in which we directly linked COT NBA or Trolox to the fluorophore through a flexible, 12-atom linker (Supplementary Fig. 1). We developed a general strategy for the synthesis of such compounds by first modifying each TSQ to contain a single, amine-functional group followed by coupling it t the commercially available, his-reactive, N-hydroxysuccinimid NHS-Cv5-TSO species at high efficiency (-30-60%) and purity (>98%) (Fig. 1a and Supplementary Note).
Bulk fluorescence measurements of the TSQ-conjugated Or

fluorophores demonstrated that absorption and emission spectr

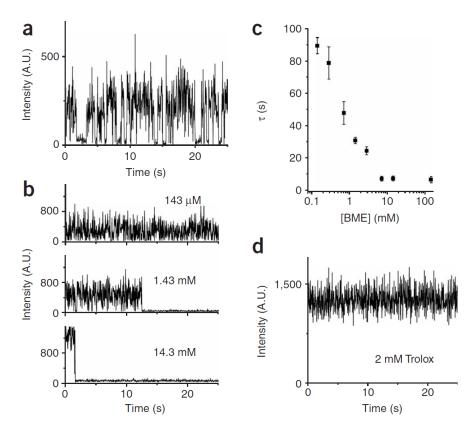
applications demanding high signal-to-noise ratios.

The addition of small-molecule solution additives is a duplex DNA oligonucleotide that we purified to homogeneity concentration-dependent manner to reduce blinking rates, dark-state lifetimes and photobleaching rates. state lifetimes and photobleaching rates.

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## **Photoblinking and photobleaching**



Rasnik et al., 2006

## **Cyanine fluorophore derivatives**

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### COT-Cy5-NHS:

LCMS: 25-65% B over 2.5 min, rt = 1.73 min

HPLC: 25-65% B over 25 min, rt = 13.01 min

ESI-MS: m/z calculated for C54H65N5O12S2

[M+H]<sup>+</sup> 1040.4, found 1040.7

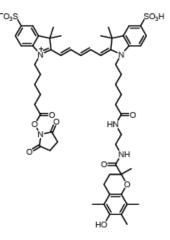
### TX-Cy5-NHS:

LCMS: 25-65% B over 2.5 min, rt = 1.80 min

HPLC: 25-65% B over 25 min, rt = 13.65 min

ESI-MS: m/z calculated for C<sub>57</sub>H<sub>71</sub>N<sub>5</sub>O<sub>14</sub>S<sub>2</sub>

[M+H]<sup>+</sup> 1114.5, found: 1114.8



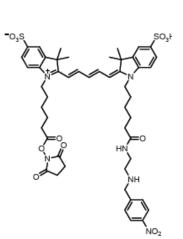
#### NBA-Cy5-NHS:

LCMS: 25-65% B over 2.5 min, rt = 1.28 min

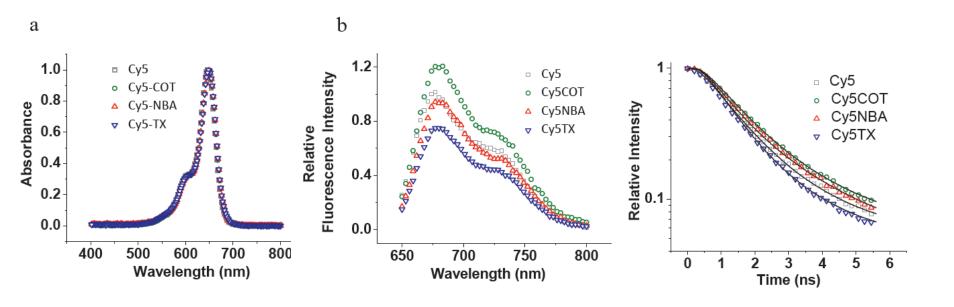
HPLC: 25-65% B over 25 min, rt = 7.33 min

ESI-MS: m/z calculated for  $C_{50}H_{60}N_6O_{13}S_2$ 

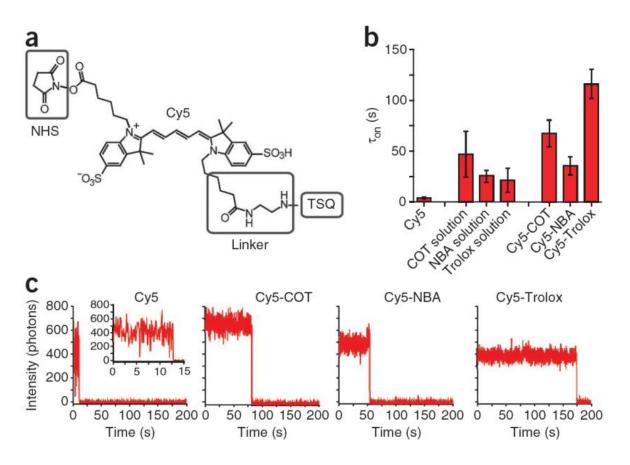
[M+H]<sup>+</sup> 1017.4, found: 1017.7



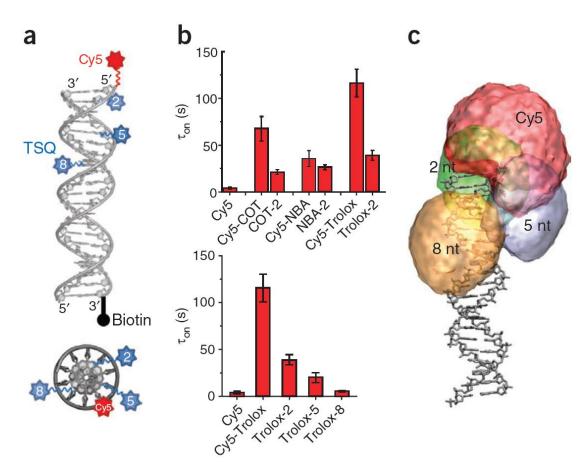
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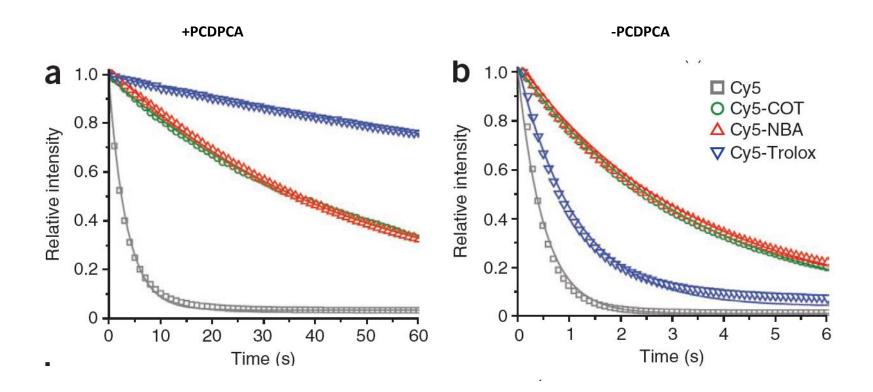
### Single molecule traces



### distance control



### With Oxygen scavenging system



# on the living cell

