## Potential-dependent structure of the interfacial water on the gold electrode

S. Nihonyanagi et al., Surface Science, 573, 11-16 (2004)

2014. 07. 19. Jonggwan Lee Soft-matter optical spectroscopy Lab. Dept. of Physics, Sogang University



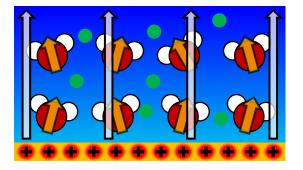
## Abstract

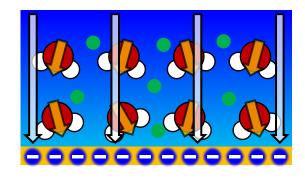
Doubly tunable sum frequency generation (SFG) spectra demonstrate that the water molecules at gold/electrolyte interface change their orientation with applied potential. At negative potentials, water molecules in the double layer align with their oxygen atom pointing to the solution. As potential became positive to be close to the potential of zero charge (PZC), the SFG signal decreased, suggesting the OH groups of the water molecule are either in random orientation or parallel to the electrode. As potential became more positive than the PZC, the SFG signal increased again with the oxygen-up orientation as same as in the negative potential region, indicating that water molecules interact with the adsorbed sulfate anions. The peak position of the SFG spectra indicates a relatively disordered state of water molecules at the gold electrode surface, in contrast to the previously observed ice-like structure of water at electrolyte/oxide interfaces.

© 2004 Elsevier B.V. All rights reserved.

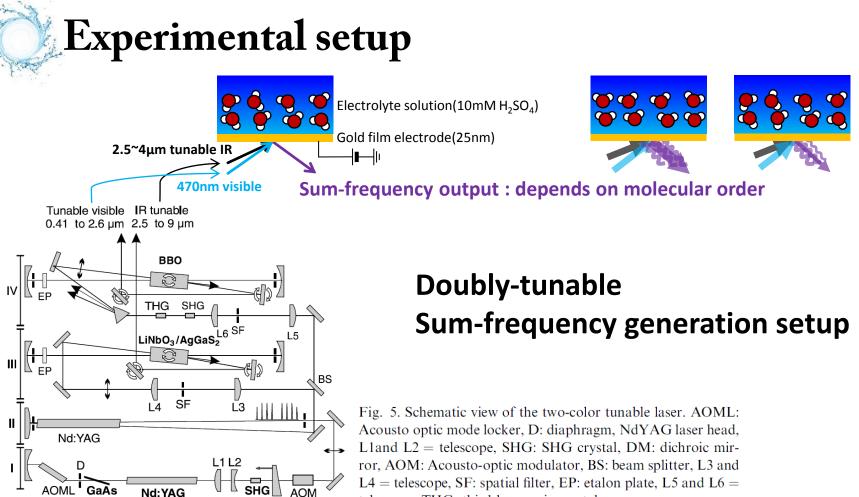
*Keywords:* Sum frequency generation; Electrochemical methods; Surface structure, morphology, roughness, and topography; Water; Metal–electrolyte interfaces





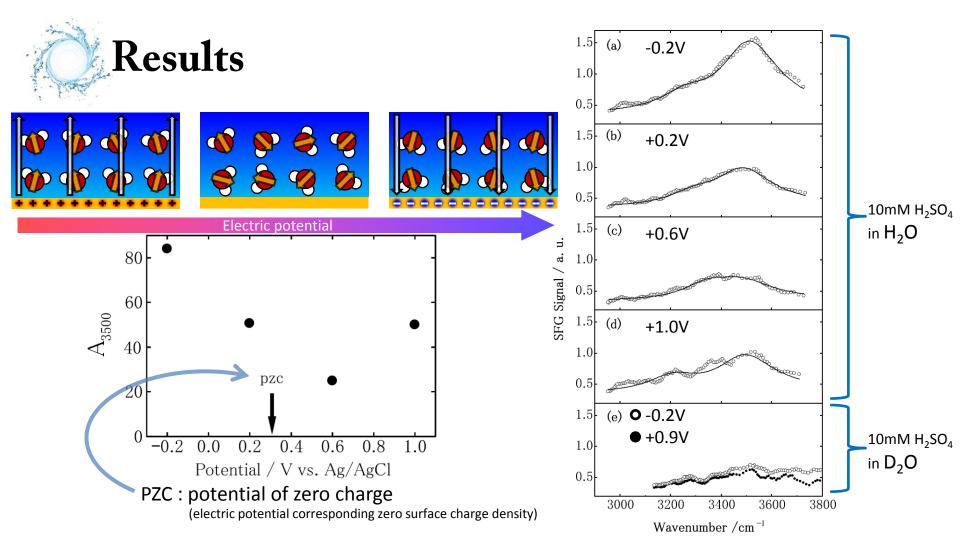


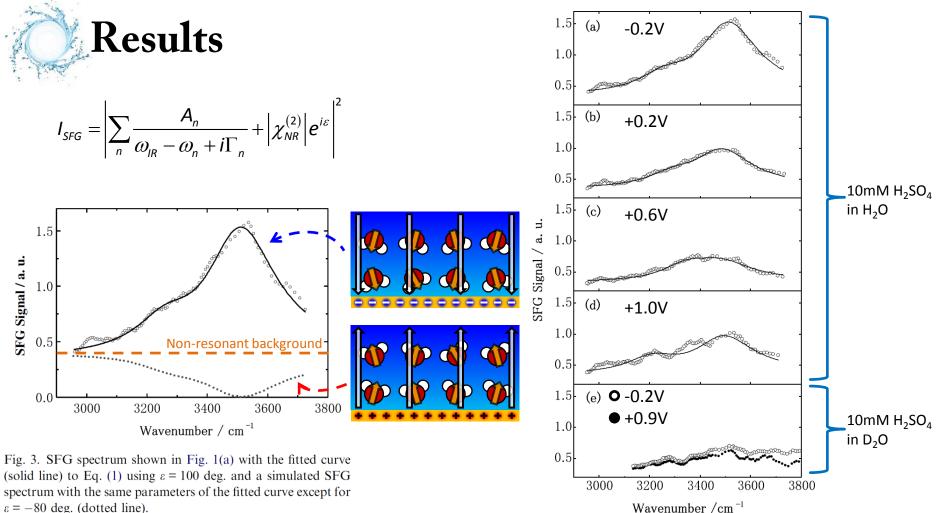
Interfacial water molecules are expected to be ordered near the charged surface, due to the electric field.



A. A. Mani et al., Surface Science 502–503 (2002) 261–267 DM

telescope, THG: third harmonic crystal.





 $\varepsilon = -80$  deg. (dotted line).

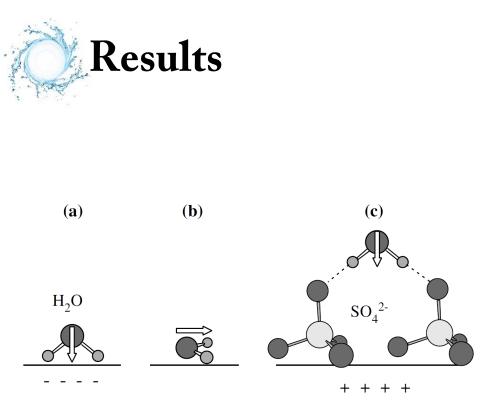


Fig. 4. Schematic models of the interfacial structure at potentials (a) more negative than, (b) around and (c) more positive than the PZC. Arrows show the direction of the water dipole.

