ELLIPSOMETRIC STUDY OF THE SURFACE OF SIMPLE LIQUIDS

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The coefficient of ellipticity $\bar{\rho}$ has been measured for liquid argon between 85 and 120 K and carbon tetrachloride between 20°C and 40°C. The experimental technique which is ideally suited to this measurement is described in detail. From $\bar{\rho}$ one is able to derive the thickness of the liquid-vapour interface. Theories of the liquid-vapour interface are reviewed and predictions compared with experiment. Theoretical uncertainties are emphasised.

$$\frac{-}{\rho} = i \frac{\pi}{\lambda} \frac{\sqrt{\varepsilon_1 + \varepsilon_2}}{\varepsilon_1 - \varepsilon_2} \cdot \int \frac{(\varepsilon - \varepsilon_1)(\varepsilon - \varepsilon_2)}{\varepsilon} dz$$

 $\frac{\varepsilon_1)(\varepsilon - \varepsilon_2)}{\varepsilon} dz$ Coefficient of Ellipticity

1

$$\left\langle \xi_{w}^{2}\right\rangle =\frac{k_{B}T}{2\pi\sigma}\ln\frac{k_{\max}}{k_{\min}}$$

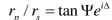
Surface Wave Excitation Theory

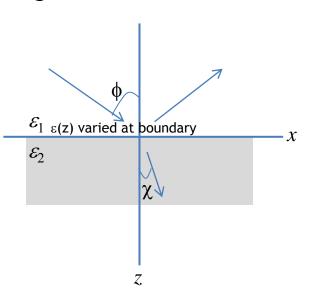
(the mean square displacement of the surface due to

(the mean square displacement of the surface due to excitation)

Seok, Sangjun -150CT2010-

Background - coefficient of ellipticity





Incidence light is plane-polarized at 45° and the Brewster angle

Snell's law

$$\sqrt{\varepsilon_1}\sin\phi = \sqrt{\varepsilon_2}\sin\chi$$

For from this it follows that

$$\varepsilon_1 \cos^2 \phi - \varepsilon_2 \cos^2 \chi = \varepsilon_1 - \varepsilon_2$$

$$\varepsilon_2 \cos^2 \phi - \varepsilon_1 \cos^2 \chi = \frac{\varepsilon_1 - \varepsilon_2}{\varepsilon_2} (\varepsilon_1 \sin^2 \phi - \varepsilon_2 \cos^2 \phi)$$

In plane wave

$$u = R_{s,p} \cos[(k \cdot r - \omega t) + \delta]$$

u is the part of the complex quantity

$$u = R_{s,p} e^{i[(k \cdot r - \omega t) + \delta]}$$

Writing now

 $u=R_{s,p}e^{i\delta}=\mathrm{R}_{\mathrm{S.P}}$ Proportional to the thickness "l"

I will be replaced by the complex amplitudes

$$\frac{R_{P}}{E_{P}} = \frac{\cos\phi\sqrt{\varepsilon_{2}} - \cos\chi\sqrt{\varepsilon_{1}}}{\cos\phi\sqrt{\varepsilon_{2}} + \cos\chi\sqrt{\varepsilon_{1}}} \left\{ I + i\frac{4\pi}{\lambda}\cos\phi\sqrt{\varepsilon_{1}} \frac{-p\cos^{2}\chi - l\varepsilon_{2}^{2}\sin^{2}\chi}{\varepsilon_{2}\cos^{2}\phi - \varepsilon_{1}\cos^{2}\chi} \right\} \dots 1$$

$$\frac{R_{s}}{E_{s}} = \frac{\cos\phi\sqrt{\varepsilon_{1}} - \cos\chi\sqrt{\varepsilon_{2}}}{\cos\phi\sqrt{\varepsilon_{1}} + \cos\chi\sqrt{\varepsilon_{2}}} \left\{ I + i\frac{4\pi}{\lambda}\cos\phi\sqrt{\varepsilon_{1}} \frac{l\varepsilon_{2} - p}{\varepsilon_{1}\cos^{2}\phi - \varepsilon_{2}\cos^{2}\chi} \right\} ...2$$

$$\frac{R_{P}}{R_{S}} = -\frac{\cos(\phi + \chi)}{\cos(\phi - \chi)} \left\{ I + i \frac{4\pi}{\lambda} \frac{\varepsilon_{2} \sqrt{\varepsilon_{1}}}{\varepsilon_{1} - \varepsilon_{2}} \cdot \frac{\cos\phi \sin^{2}\phi}{\varepsilon_{1} \sin^{2}\phi - \varepsilon_{2} \cos^{2}\phi} \eta \right\}$$

$$\eta = p - l(\varepsilon_1 + \varepsilon_2) + q\varepsilon_1\varepsilon_2$$

At the Brewster angle

$$\frac{R_{P}}{R_{S}} = i \frac{\pi}{\lambda} \frac{\sqrt{\varepsilon_{1} + \varepsilon_{2}}}{\varepsilon_{1} - \varepsilon_{2}} \eta$$

Background - coefficient of ellipticity

$$\frac{R_{P}}{R_{S}} = i \frac{\pi}{\lambda} \frac{\sqrt{\varepsilon_{1} + \varepsilon_{2}}}{\varepsilon_{1} - \varepsilon_{2}} \eta \qquad \eta = p - l(\varepsilon_{1} + \varepsilon_{2}) + q\varepsilon_{1}\varepsilon_{2} \qquad \int_{1}^{2} dz = l, \int_{1}^{2} \varepsilon dz = p, \int_{1}^{2} \frac{1}{\varepsilon} dz = q$$

$$R_{P} = R_{p} \cdot e^{i\delta_{p}}, R_{P} = R_{s} \cdot e^{i\delta_{s}}$$
 $\frac{R_{P}}{R_{S}} = \frac{R_{P}}{R_{S}} e^{i(\delta_{p} - \delta_{s})} = \rho \cdot e^{i\Delta}$

 ρ is the amplitudes and Δ the difference in phase of the two components.

$$-\frac{1}{\rho} = i\frac{\pi}{\lambda} \frac{\sqrt{\varepsilon_1 + \varepsilon_2}}{\varepsilon_1 - \varepsilon_2} \eta, \ \Delta = \pi/2$$

$$\frac{-}{\rho} = i \frac{\pi}{\lambda} \frac{\sqrt{\varepsilon_1 + \varepsilon_2}}{\varepsilon - \varepsilon} \cdot \int \frac{(\varepsilon - \varepsilon_1)(\varepsilon - \varepsilon_2)}{\varepsilon} dz$$
 Coefficient of ellipticity
$$\frac{-}{\rho_{21}} = -\frac{-}{\rho_{12}}$$

Ellipsometry techniques

If the angle of incidence is set to θ_B when $\Delta=\pi/2$ (by adjusting θ until Re(r)=0) then includes residual static phase shift has only a small effect on Im(r)

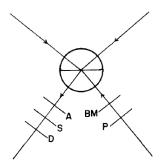


Fig. 1. The optical arrangement. Light beams reflected from the liquid—vapour and vapour—liquid surfaces are shown. P Polariser, BM birefringence modulator; A analyser; S horizontal slit; D detector.

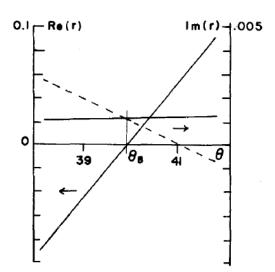


Fig. 2. Calculated θ variation of r_{12} for a uniform film (ϵ = 1.22, t = 15 A) on a bulk medium (ϵ_2 = 1.5). The dashed line shows Im(r) when a stray Δ_1 of 0.02 rad is present. The Re(r) is not affected on this scale.

Results - Carbon tetrachloride

Table I Summary of data for carbon tetrachloride

Constar	its d	= 5.16 Å	(45% pac	king f	raction)								
		σ (erg	/cm ²) ^a	ϵ_1^{b})	ϵ_2^{b}		$\tilde{\rho} \times 10^4/\eta$		$\eta/t(F)$ A^{-1}	η/t (erg A ⁻¹)	η/k _{max} (Å ^c)	
	8°C 10°C	27 24			011 017	2.132 2.102		-7.77 -7.95		0.194 0.186	-0.188 -0.180	-1.88 -2.15	
Experin	nent								from the		surface exci	tation theory	$k_{\rm max} = 2\pi/t$
	₽ ₁₂ ×	104	₽ ₂₁ ×	10 ⁴	א קו × 10	0 ⁴ 1 _{av}	t(F) (Å)	t(Erf) (A)	k _{max} (Å)-1	t _м (Å)			
18°C	11.8 ±	0.3	-12.9 ±	0.3	12.3 ±	0.3	8.2	8.5	0.85	7.4	$\bar{\rho}_{21}(40^{\circ}\text{C})/\bar{\rho}_{21}(1$	8°C) = 1.10 ± 0.04	
Other v	vork										$\frac{\overline{\rho}_{21}}{\overline{\rho}_{12}} = 1.09$	3 ± 0.05 at 18	, C
Referen	nce	⊅ ₁₂ ×	104	T°C							This differe	nce points to	an inadequacy
3 8.4 4 12.6 $\pm \frac{1}{2}$ 5 10.5 $\pm \frac{1}{2}$ 6 17.8 $\pm 1\frac{1}{2}$		Room temperature 12½ 15-18 no T dependence to 140°C 20, p rising to 26 at 40°C							This difference points to an inadec of the assumption of the Drude mo				

Space averaged density profile (error fn form)

$$\rho_{\rm Erf}(z) = \frac{\rho_l + \rho_v}{2} + \frac{\rho_l - \rho_v}{2} \, {\rm Erf}\left(\frac{z}{\sqrt{2} \xi_{\rm rms}}\right).$$

Fermi profile
$$\eta(F) = (\epsilon_2 - \epsilon_1) \ln \frac{\epsilon_1}{\epsilon_2} \delta$$

a International Critical Tables, vol. 4, p. 447.
 b Interpolated from International Critical Tables, vol. 7, p. 12.

^c Using σ experimental.

Results - Liquid argon

The reservoir was cooled to 77 K using liquid nitrogen

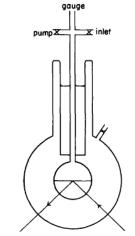


Fig. 3. The pyrex glass cell used for the liquid argon studies.

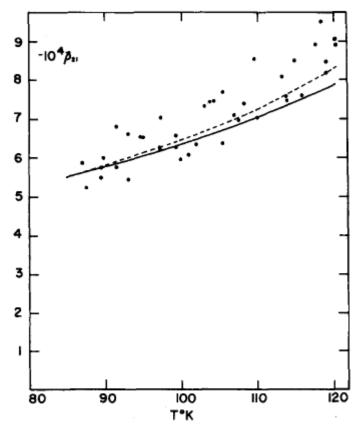


Fig. 4. p_{21} measured for liquid argon in 3 warming cycles between 85 K and 120 K. The full line shows the variation of p predicted by the surface wave excitation theory with $k_{\min} = 2\pi/\lambda$, $k_{\max} = 2\pi/t$, using the experimental value of surface tension, while the dashed line holds k_{\max} constant.

The mean square displacement

$$\langle \xi_{\rm w}^2 \rangle = \frac{k_{\rm B}T}{2\pi\sigma} \ln \frac{k_{\rm max}}{k_{\rm min}}$$

Results - Liquid argon

$$e(z) = 1 + \frac{n\alpha/\epsilon_0}{1 - n\alpha/3\epsilon_0}$$

From the Clausius-Mossotti

Table II Summary of liquid argon data

Density variation

					1							
Const	tants d	3.40 Å, 7	r _c = 150.9	K _		Su	rface	tensio	n			
T (K)	$\epsilon_1^{\ a}$	1 ^a ε2 ^a θΒ (°)		ρ_l^b (g/cm ³)	ρ _ν b (g/cm ³)	σ ^c (erg/cm ²)		ρ /η		$\eta/t(F)$	η/t(Erf)	η/k _{max} (A ²)
85	1.0015	1.5147	39.11	1.402	0.0046	13.12		-1.535×10^{-3}		-0.0483	-0.0467	-0.281
90	1.0025	1.5026	39.24	1.374	0.0080	11.86		1.571		0.0461	0.0445	0.314
100	1.0056	1.4750	39.55	1.309	0.0180	9.42		1.668		0.0409	0.0395	0.390
110	1.0103	1.4450	39.90	1.238	0.0328	7.10		1.787		0.0354	0.0343	0.493
120	1.0183	1.4137	40.32	1.160	0.0580	4.95		1.961		0.0295	0.0285	0.645
Expe	riments				$\eta(F) = (\epsilon$	$\epsilon_2 - \epsilon_1$) h	$n \frac{\epsilon_1}{\epsilon_2} \delta$	$ \rho_{\text{Erf}}(z) = \frac{\rho}{1} $	$\frac{l+\rho_v}{2}+\frac{\rho_l}{2}$	$\frac{1-\rho_{v}}{2}$ Erf $\left(\frac{1}{2}\right)$	$\frac{z}{\sqrt{2}\xi_{\text{rms}}}$). \overline{k}_{n}	$\frac{1}{1} = 2\pi / t$
T	-> ₂₁ ×	104		-η		t(F)	-2 -	t(Etf)		kmax		t _m
(K)	e	f		(A)		(A)		(A)		(A-1)		(Å)
85	4.7 ± 0.	.4 (5.4	± 0.4)	0.306	(0.352)	6.33	(7.30)	6.55	(7.54)	1.09	(1.25)	5.8 (5.0)
90		5.7 ± 0.4			0.363		7.9		8.2		1.16	5.4
100		6.5 ± 0.4			0.390		9.5		9.9		1.00	
110		7.5	± 0.4		0.420		11.9	12.2			0.85	
120		8.8	± 0.6		0.450	15.2 15.8			15.8		0.70	9.0
Predic	ctions		$\sigma_0 = \sigma$	$a + \frac{3}{16\pi} k_{\rm B}$	Tk2max.	Tempera	ature var	riation				
T (K)	tw ^g (A)	twh (A)	σ ₀ (erg/cm ²	t_{wo}^{i}	t _i ^j (A)	$t_m(T)/t$	_m (90) ^k	t(Erf, 7	/)/t(Erf, 9	90) ^k t _w	(T)/t _w (90)	l(T)/l(90)
85	7.64	7.59	20.52	6.22	4.3	0.93		0.92		0.9	93	0.95
90	8.22	8.22	18.78	6.67		1		1	1		1	
100	9.62	9.72	15.38	7.70	6.2	1.17		1.21		1.17		1.12
110	11.47	11.74	11.87	9.09	8.1	1.37		1.49		1.39		1.29
120	14.12	14.69	8.57	11.03	11.3	1.67		1.93		1.7	72	1.54

^a Extrapolated from the data of Sinnock and Smith, ref. 29 using the Clausius-Mossotti expression. ^b Ref. 30. ^c Ref. 31. ^d Using σ experimental. ^e Average of 3 samples. ^f Average of 3 warming cycles, value in parentheses extrapolated to 85 K. ^g $k_{\min} = 2\pi/\lambda$, $k_{\max} = 2\pi/t$, σ experimental. ^h $k_{\min} = 2\pi/\lambda$, k_{\max} constant = $2\pi/t$ (90), σ experimental. ⁱ $k_{\min} = 2\pi/\lambda$, $k_{\max} = 2\pi/t$, σ_0 bare. ^j $t_1^2 = t^2 (\text{Erf}) - t_{\text{w0}}^2$. Experimental values.